BIRLA INSTITUTE OF TECHNOLOGY, MESRA, RANCHI (END SEMESTER EXAMINATION)

CLASS: MSc/IMSc/Pre_PhD SEMESTER: VIII/ II/ PrePhD

BRANCH: Chemistry SESSION: SP/2024

SUBJECT: CH412 ANALYTICAL CHEMISTRY

TIME: 3 Hours FULL MARKS: 50

INSTRUCTIONS:

- 1. The question paper contains 5 questions each of 10 marks and total 50 marks.
- 2. Attempt all questions.
- 3. The missing data, if any, may be assumed suitably.
- 4. Before attempting the question paper, be sure that you have got the correct question paper.
- 5. Tables/Data hand book/Graph paper etc. to be supplied to the candidates in the examination hall.

Q.1(a)	Compare and contrast population and sample standard deviation with relevant	[5]	CO 1	BL 2
Q.1(b)	mathematical expressions. Which among the two is (a) more accurate (b) more practical. Outline the concept of <i>sample spiking technique</i> of third-party inspection. Elaborate with example of your choice.	[5]	1	3
Q.2(a)	Compare and contrast tubular column and capillary column of GC. Highlight their advantages and disadvantages.	[5]	2	2
Q.2(b)	Explain with clear illustration the concept of 2D TLC. What is its advantage over normal TLC?	[5]	2	2
Q.3(a)	Outline the <i>Volhard's method</i> of estimation of chloride in a given sample solution. Discuss the importance of chloride estimation in industrial waters.	[5]	3	2
Q.3(b)	Outline the gravimetric estimation of calcium in water samples. Discuss in details the chemistry behind the estimation method.	[5]	3	2
Q.4(a) Q.4(b)	Draw a TGA curve for Calcium oxalate or any other salt and discuss its interpretation? Classify the types of thermal analysis, discuss the role of DSC, TGA and DTA giving any one example each?	[5] [5]	4	2 2
Q.5(a)	Outline the Potentiometric-Argentometric estimation of chlorides in a given water sample. Draw all the relevant schematics and chemical equations.	[5]	4	2
Q.5(b)	Discuss the various types of 'single electrodes'. Elaborate the application of metal-sparingly soluble salt type single electrode as reference electrode.	[5]	4	2

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