

BIRLA INSTITUTE OF TECHNOLOGY, MESRA, RANCHI
(END SEMESTER EXAMINATION)

CLASS: IMSc
BRANCH: CHEMISTRY

SEMESTER : VI
SESSION : SP/2024

SUBJECT: CH313R1 INORGANIC CHEMISTRY-IV

TIME: 3 Hours

FULL MARKS: 50

INSTRUCTIONS:

1. The question paper contains 5 questions each of 10 marks and total 50 marks.
 2. Attempt all questions.
 3. The missing data, if any, may be assumed suitably.
 4. Before attempting the question paper, be sure that you have got the correct question paper.
 5. Tables/Data hand book/Graph paper etc. to be supplied to the candidates in the examination hall.
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		CO	BL
Q.1(a)	What is Collman's reagent? Discuss its synthesis and applications in organic synthesis.	[4]	1 2
Q.1(b)	Most of the organometallic carbonyl complexes undergo migrations, not insertion. How migration is different from insertion? Discuss the migration vs insertion in $\text{CH}_3\text{Mn}(\text{CO})_5$.	[4]	1 3
Q.1(c)	What is β -hydride elimination? Which of the following metal-alkene complex is most unstable and why? a. $\text{Ti}(\text{C}_2\text{H}_4)_4$ b. $\text{Ti}(\text{CH}_2\text{Ph})_4$ c. $\text{Pb}(\text{CH}_3)_4$ d. $\text{W}(\text{CH}=\text{C}(\text{CH}_3)_2)_6$	[2]	1 2
Q.2(a)	Discuss metal-(1,3-Butadiene) bonding based on Molecular orbital theory and discuss synergistic effect. what are the different binding modes possible in metal-alkene and metal-alkyne complexes?	[5]	2 2
Q.2(b)	Discuss the bonding of 1,4-butadiene with transition metal using the molecular orbital approach.	[3]	2 1
Q.2(c)	Why alkenes bound to metal are prone to nucleophilic attack rather than electrophilic attack?	[2]	2 1
Q.3(a)	What is oxidative coupling? Explain the oxidative coupling in $\text{Ir}(\text{PPh}_3)_3\text{Cl}$.	[4]	3 2
Q.3(b)	Discuss the oxidative addition of a) Cl_2 , b) O_2 , and c) methyl iodide (MeI) in Vaska's complex	[4]	3 3
Q.3(c)	What is Schlenk equilibrium?	[2]	3 1
Q.4(a)	Explain the trans effect with examples.	[5]	4 1
Q.4(b)	Discuss the mechanism of electrostatic polarization and pi bonding theory to explain the trans effect.	[5]	4 2
Q.5(a)	What are the differences between homogeneous and heterogeneous catalysis? Define Turnover number (TON) and Turnover frequency (TOF) in catalysis reactions.	[5]	5 2
Q.5(b)	Discuss catalytic cycles for the following - 1. Wacker-oxidation of alkene 2. Monsanto process for acetic acid synthesis by catalytic carbonylation of methanol	[5]	5 3

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