



Name: ..... Roll No.: .....

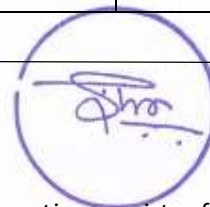
Branch: ..... Signature of Invigilator: .....

Semester: IVth Date: 27/04/2022 (MORNING)

Subject with Code: CH207 INORGANIC CHEMISTRY-III (COORDINATION CHEMISTRY)

Marks Obtained	Section A (30)	Section B (20)	Total Marks (50)

INSTRUCTION TO CANDIDATE



1. The booklet (question paper cum answer sheet) consists of two sections. First section consists of MCQs of 30 marks. Candidates may mark the correct answer in the space provided / may also write answers in the answer sheet provided. The Second section of question paper consists of subjective questions of 20 marks. The candidates may write the answers for these questions in the answer sheets provided with the question booklet.
2. The booklet will be distributed to the candidates before 05 minutes of the examination. Candidates should write their roll no. in each page of the booklet.
3. Place the Student ID card, Registration Slip and No Dues Clearance (if applicable) on your desk. All the entries on the cover page must be filled at the specified space.
4. Carrying or using of mobile phone / any electronic gadgets (except regular scientific calculator)/chits are strictly prohibited inside the examination hall as it comes under the category of unfair means.
5. No candidate should be allowed to enter the examination hall later than 10 minutes after the commencement of examination. Candidates are not allowed to go out of the examination hall/room during the first 30 minutes and last 10 minutes of the examination.
6. Write on both side of the leaf and use pens with same ink.
7. The medium of examination is English. Answer book written in language other than English is liable to be rejected.
8. All attached sheets such as graph papers, drawing sheets etc. should be properly folded to the size of the answer book and tagged with the answer book by the candidate at least 05 minutes before the end of examination.
9. The door of examination hall will be closed 10 minutes before the end of examination. Do not leave the examination hall until the invigilators instruct you to do so.
10. Always maintain the highest level of integrity. Remember you are a BITian.
11. Candidates need to submit the question paper cum answer sheets before leaving the examination hall.



**Short answer type questions**

**Answer any five**

**5x4 = 20 marks**

1. Validate the postulates of Warner's theory for the complex  $[\text{Co}(\text{NH}_3)_5\text{Cl}]\text{Cl}_2$ . 4
2. Identify the following as Inner sphere or outer sphere complex:  $\text{Fe}(\text{CN})_6^{4-}$ ,  $[\text{CoF}_6]^{3-}$ . 4
3. Draw the orbital splitting pattern of  $Z_{\text{in}}$  and  $Z_{\text{out}}$  type of Jahn Teller distortion. 4
4. Draw the molecular orbital diagram of  $[\text{Ti}(\text{H}_2\text{O})_6]^{3+}$ . 4
5. Discuss the structure, differences, and oxygen transport process by Hemoglobin and Myoglobin. 4
6. a) Considering the spin-orbit coupling in lanthanide elements, calculate the magnetic moment of  $\text{Ln}^{3+}$  systems which contains i) four electrons in f-orbitals ( $f^4$ -system) and ii) 12 electrons in f-orbitals ( $f^{12}$ -system). 2
- b) Why lanthanides are characterized by a uniform (+III) oxidation state? 2