BIRLA INSTITUTE OF TECHNOLOGY, MESRA, RANCHI (END SEMESTER EXAMINATION)

CLASS: IMSc SEMESTER: V
BRANCH: CHEMISTRY SESSION: MO/2024

SUBJECT: CH327 ORGANIC CHEMISTRY-IV

TIME: 3 Hours FULL MARKS: 50

INSTRUCTIONS:

- 1. The question paper contains 5 questions each of 10 marks and total 50 marks.
- 2. Attempt all questions.
- 3. The missing data, if any, may be assumed suitably.

Q.5(b) Draw Hantzsch's synthesis of Pyrrole with the mechanism.

- 4. Before attempting the question paper, be sure that you have got the correct question paper.
- 5. Tables/Data hand book/Graph paper etc. to be supplied to the candidates in the examination hall.

CO BL Discuss the fundamental equation of NMR and explain the term Chemical Shift. How is it [5] 1 measured relative to Tetramethyl Silane? Q.1(b) Explain with reason if you can differentiate ethane, ethylene and acetylene by NMR [5] 1 Spectroscopy. Q.2(a) Indicate what NMR spectra would you expect for the following compounds: i) CH₂CH₂CH₂CH₃ [5] 2 2 ii) (CH₃)4C iii) 1,2,2-Trichloropropane iv) Toluene An organic compound with a molecular formula, $C_{10}H_{12}O_2$ gave an IR band (C=O stretching) [5] 2 3 at 1740 cm-1. The NMR shows a triplet at δ = 1.2 (3H); singlet at δ =3.5(2H), quartet δ =4.1 (2H) and a multiplet at δ =7.3 (5H). Q.3(a) i) Convert D-Glucose Fisher Form to Pyranose Haworth Form. Draw and distinguish the a [5] 3 4 and B-form. ii) What is Mutarotation? Q.3(b)i) Draw the osazone formation mechanism for D-glucose? ii) Demonstrate the epimerization [5] 3 mechanism. Q.4(a) What is Pericyclic Reaction. Explain with example reaction with classification. 1 Draw the correlation diagram for the following thermal electrocyclic ring closure and [5] 3 Q.4(b) explain the possible stereochemical product formation. (Thermal) Ме Ме Q.5(a) Draw a molecular orbital diagram and demonstrate the resonance structure of furan, 2 [5] 1 pyrrole, and pyridine. Compare the reactivity.

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[5] 2