BIRLA INSTITUTE OF TECHNOLOGY, MESRA, RANCHI (END SEMESTER EXAMINATION)

CLASS: B.PHARM SEMESTER: III
BRANCH: PHARMACY SESSION:MO-2024

SUBJECT: BP302 T PHYSICAL PHARMACEUTICS

TIME: 3.00 Hours FULL MARK: 75

INSTRUCTIONS:

- 1. The missing data, if any, may be assumed suitably.
- 2. Before attempting the question paper, be sure that you have got the correct question paper.
- 3. Tables/Data hand book/Graph paper etc. to be supplied to the candidates in the examination hall.
- 4. This question paper consists of (03) three parts. Read the part wise instructions before attempting the questions.

PART-I

Objective types of questions (Instruction: Answer all questions)

Q1. $(10 \times 2 = 20 \text{ Marks})$

- A. What is Buffer Capacity?
- B. Explain the process of liquefaction of gases.
- C. State the limitations of Raoult's Law.
- D. Define Vapour pressure.
- E. Define Surface tension and Interfacial Tension.
- F. What are clathrates?
- G. Define adsorption isotherm.
- H. Explain the term detergency
- I. What are aerosols? Give two examples of aerosols

PART-II

Short Answers

(Instruction: Answer seven out of nine questions)

 $(7 \times 5 = 35 \text{ Marks})$

- Q2. Explain critical solution temperature and mention its applications.
- Q3. Define dielectric constant. Write a note on its application in pharmacy.
- Q4. Write down the calorimetric method of pH determination.
- Q5. Explain the solubility of partially miscible liquids with the help of a suitable example.
- Q6. Discuss Langmuir adsorption isotherm.
- Q7. Discuss Protein binding with the help of examples.
- Q8. What is the mechanism of cyclodextrin drug inclusion complexes? Discuss its applications.
- Q9. What are eutectic mixtures. Give two examples.
- Q10. Write a note on the application of buffers in pharmaceutical and biological systems.

PART-III

Long Answers

(Instruction: Answer two out of three questions)

 $(2 \times 10 = 20 \text{ marks})$

- Q11. Enlist the various physical properties of drug molecules. Discuss any two of them with suitable example.
- Q12. Define Surface tension. Explain the principle involved in determination of surface tension by capillary rise method.
- Q13. What do you mean by complexation? Enlist the various methods used for analysing complexes. Explain in detail the distribution method of complex analysis.

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