

BIRLA INSTITUTE OF TECHNOLOGY, MESRA, RANCHI  
(END SEMESTER EXAMINATION)

CLASS: IMSc  
BRANCH: IMSC CHEMISTRY

SEMESTER : III  
SESSION : MO/2023

SUBJECT: CH201R1 INORGANIC CHEMISTRY-II

TIME: 3 Hours

FULL MARKS: 50

**INSTRUCTIONS:**

1. The question paper contains 5 questions each of 10 marks and total 50 marks.
2. Attempt all questions.
3. The missing data, if any, may be assumed suitably.
4. Before attempting the question paper, be sure that you have got the correct question paper.
5. Tables/Data hand book/Graph paper etc. to be supplied to the candidates in the examination hall.

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| Q.1(a) Outline the principles of refining metals by the following methods:<br><div style="margin-left: 40px;">i) Refining of Titanium by electrolytic Kroll Process</div> <div style="margin-left: 40px;">ii) Vapour Phase refining of Nickel and Zirconium</div>  | [5] | 1 2 |
| Q.1(b) (i) Outline the principles of refining metals by a) Zone refining and b) Mond's Process.<br>(ii) Explain the Hall-Heroult process for the isolation of Aluminium metal.   | [5] | 1 2 |
| Q.2(a) (i) Compare the aromaticity and total valence electron counts (TVE) of Benzene and boroxine.<br>(ii) Write the reactions for a) complete and ii) partial hydrolysis of borazine.  | [5] | 2 3 |
| Q.2(b) (i) Explain the solvent system definition of acid and base. Write the auto-ionization of BrF <sub>3</sub> and predict whether BrF <sub>2</sub> AsF <sub>6</sub> is acid or base if it is soluble in BrF <sub>3</sub> solvent.<br>(ii) How aqua acid is different from hydroxoacids and oxoacids? Give one example in each case.   | [5] | 2 2 |
| Q.3(a) What do you mean by inorganic benzene? Draw the structure of Basic beryllium acetate.   | [5] | 3 2 |
| Q.3(b) Discuss the structure and properties of the allotropes of carbon.   | [5] | 3 1 |
| Q.4(a) What are pseudohalogens? Discuss the structure of IF <sub>7</sub> and ClF <sub>3</sub> .  | [5] | 4 2 |
| Q.4(b) Explain the preparation, properties and structure of XeF <sub>6</sub> .   | [5] | 4 1 |
| Q.5(a) Identify the structure of x <sub>1</sub> , y <sub>1</sub> , x <sub>2</sub> , and y <sub>2</sub> in the following reactions.<br><div style="margin-left: 40px;">(i) <math>3\text{B}(\text{CH}_3)_3 + 3\text{NH}_3 \xrightarrow[20 \text{ atm}]{320-340^\circ\text{C}} \text{x}_1 + 6\text{CH}_4</math></div> <div style="margin-left: 80px;"><math>\text{x}_1 + 3\text{H}_2\text{O} \longrightarrow \text{y}_1 + 3\text{NH}_3</math></div> <div style="margin-left: 40px;">(ii) <math>3\text{B}_2\text{H}_6 + 6\text{NH}_2\text{CH}_3 \xrightarrow{180-200^\circ\text{C}} \text{x}_2 + 12\text{H}_2</math></div> <div style="margin-left: 80px;"><math>\text{x}_2 + 3\text{H}_2\text{O} \xrightarrow{60^\circ\text{C}} \text{y}_2 + 3\text{H}_2</math></div> | [5] | 5 3 |
| Q.5(b) Write the reactions for the following transformation and identify the structure of reactants and product/(s)<br><div style="margin-left: 40px;">i) Hydrolysis of <i>cyclic-tetra</i>-phosphonitrilic chloride</div> <div style="margin-left: 40px;">ii) Ammonolysis of <i>cyclic-tri</i>-phosphonitrilic chloride</div> <div style="margin-left: 40px;">iii) The reaction of <i>cyclic-tri</i>-phosphonitrilic chloride with 1,2-dihydroxybenzene</div>   | [5] | 5 3 |

::: 21/11/2023 :::