

**BIRLA INSTITUTE OF TECHNOLOGY, MESRA, RANCHI  
(END SEMESTER EXAMINATION MO2022)**

CLASS: IMSC  
BRANCH: PHYSICS

SEMESTER : V  
SESSION : MO /2022

SUBJECT: PH301 QUANTUM MECHANICS AND APPLICATIONS

TIME: 03 HOURS

FULL MARKS: 50

**INSTRUCTIONS:**

1. The question paper contains 5 questions each of 10 marks and total 50 marks.
2. Attempt all questions.
3. The missing data, if any, may be assumed suitably.
4. Tables/Data handbook/Graph paper etc., if applicable, will be supplied to the candidates

- Q.1(a) What do you understand by Wave-particle duality? [2]
- Q.1(b) Calculate the wavelength associated with an electron subjected to a potential difference of 1.25 kV? [3]
- Q.1(c) A wave function  $\psi(x) = A_n \sin \frac{2n\pi x}{L}$  in the region  $0 \leq x \leq L$ . Find the value of  $A_n$  using normalization condition? [5]
- Q.2(a) Write down the time independent and time-dependent Schrodinger equation? [2]
- Q.2(b) For a rectangular potential barrier of width  $a$  and height  $V_0$ , show that the transmission co-efficient for ( $E > V_0$ ) can be expressed as  $T = \frac{4p_1^2 p_2^2}{(p_1^2 - p_2^2)^2 \sin^2(p_2 a / \hbar) + 4p_1^2 p_2^2}$ . [3]
- Q.2(c) Calculate the probability of transmission of a-particle through the rectangular barrier indicated below:  $V_0 = 2$  eV,  $E = 1$  eV and barrier width = 1 Å, mass of a-particle =  $6.4 \times 10^{-27}$  kg [5]
- (use  $T = \frac{-4p_1^2 p_2^2 \sec^2 \hbar^2(ip_2 a / \hbar)}{(p_1^2 + p_2^2)^2 \tanh^2(ip_2 a / \hbar) - 4p_1^2 p_2^2}$  and apply approximation for the thick barrier  $ip_2 a \gg \hbar$ )
- Q.3(a) What do you understand by zero-point energy of a quantum oscillator? [2]
- Q.3(b) For a particle in a box show that allowed values of total energy are given by [3]
- $$E = E_x + E_y + E_z = \frac{\hbar^2}{2m} \left[ \frac{n_x^2}{l_x^2} + \frac{n_y^2}{l_y^2} + \frac{n_z^2}{l_z^2} \right]$$
- Q.3(c) Calculate the energy difference between the ground state and the first excited state for an electron in one dimensional rigid box of length  $10^{-10}$  m (mass of electron =  $9.1 \times 10^{-31}$  Kg and  $h = 6.626 \times 10^{-34}$  Joule-sec. [5]
- Q.4(a) What do you understand by de-Broglie waves? [2]
- Q.4(b) For the radial part of Hydrogen atom given by [3]
- $$\frac{\partial^2 R}{\partial r^2} + \frac{2}{r} \frac{\partial R}{\partial r} + \left[ \frac{-l(l+1)}{r^2} + \frac{2\mu}{\hbar^2} \{E - V(r)\} \right] R = 0$$
- Setup the recursion relation
- $$a_{k+1} = \frac{k+l+1-\lambda}{(k+1)(2l+k+2)} a_k$$
- Use substitutions  $\alpha^2 = -\frac{2\mu E}{\hbar^2}$  and  $\lambda = \frac{\mu Z e^2}{\hbar^2 \alpha}$
- Q.4(c) Show that energy Eigen-values are given by [5]
- $$E_n = -\frac{\mu Z^2 e^4}{2\hbar^2 n^2}$$
- Q.5(a) What do you understand by spin angular momentum of an electron? [2]
- Q.5(b) Define magnetic moment of an atom and Lande's g-factor? [3]
- Q.5(c) Mathematically show that magnetic dipole moment and orbital angular momentum of an electron are related by  $\mu_l = -\frac{e}{2m} p_l$ . Numerically calculate the value of Bohr-magneton. [5]