BIRLA INSTITUTE OF TECHNOLOGY, MESRA, RANCHI (END SEMESTER EXAMINATION)

IMSC CHEMISTRY	SEMESTER : V SESSION : MO/19
SUBJECT: IMC5003 ORGANIC CHEMISTRY-I 3 HOURS	FULL MARKS: 60
<ul> <li>INSTRUCTIONS:</li> <li>1. The question paper contains 7 questions each of 12 marks and total 84 marks.</li> <li>2. Candidates may attempt any 5 questions maximum of 60 marks.</li> <li>3. The missing data, if any, may be assumed suitably.</li> <li>4. Before attempting the question paper, be sure that you have got the correct question paper.</li> <li>5. Tables/Data hand book/Graph paper etc. to be supplied to the candidates in the examination hall.</li> </ul>	
Draw the structure of A and B.	[2]
i) $(H) \xrightarrow{HaOH, CaO} A$ ii) $(H) \xrightarrow{Zn/Hg} B$	
Discuss Baeyer strain theory. Draw the structure of A and B with possible mechanism of Simon-Smith cyclopropa	[4] nation. [6]
$H_3C \subset CH_2 + A \xrightarrow{Zn(Cu) / Ether} B + ZnI_2$	
Write down the products obtained from the reaction of HBr with 1,3-Butadiene	and discuss the [2]
Discuss the B-elimination reaction, explain the fact about major (A) and minor (B) p with mechanistic details as per following reaction $\bigcirc$	roduct formation [4]
$ \begin{array}{c} & \overset{CH_3}{\longrightarrow} & \overset{OH}{\longrightarrow} & A & + & B \\ & & & & & \\ \oplus & & & & & \\ \end{array} $	
Discuss the mechanisms of Oxymercuration-Reduction for the hydration of propene	e. [6]
Write the product obtained from the following reaction. $\begin{array}{c} & & \\ & $	[2]
Show the mechanism for the following conversion.	[4]
Write the structure of A and B. Discuss the mechanism for formation of A. $\bigcirc$ $-OH \xrightarrow{NaOH, CO_2} A \xrightarrow{H_2SO_4} B$	[6]
Discuss the mechanism of $S_N 2$ reaction with suitable example. Write down stepwise mechanism for the following reaction. $Cl_2, FeCl_3$ Base (:B)	[2] [4]
Describe the stepwise mechanism involved in the following aromatic nucleop reactions.	nilic substitution [6]
i) $\swarrow$ $N_2CI$ $\xrightarrow{CuCl}$ ii) $H_3C$ $\sim$ $CH_2$ $\xrightarrow{HBr}$ mechanism	
	CHEMISTRY SUBJECT: IMC5003 ORGANIC CHEMISTRY-I 3 HOURS TIONS: Jestion paper contains 7 questions each of 12 marks and total 84 marks. Jates may attempt any 5 questions maximum of 60 marks. Jising data, if any, may be assumed suitably. The attempting the question paper, be sure that you have got the correct question ////////////////////////////////////

Q.5(a) Complete the conversion with reagents and intermediate. [2]

$$CH_3CH_2Br \longrightarrow CH_3CH_2CH_2OH$$
  
acol-pinacolone rearrangement with mechanism. [4]

[4]

[6]

Q.5(b) Draw and discuss the pinacol-pinacolone rearrangement with mechanism. Q.5(c) Identify the product A intermediate B & final product C. Explain the steps with mechanism

$$\begin{array}{c} & \xrightarrow{OsO_4} & A \xrightarrow{HIO_4} & [B] \longrightarrow C + HIO_3 + H_2O \\ CH_3 & \xrightarrow{OsO_4} & A \xrightarrow{HIO_4} & [B] \longrightarrow C + HIO_3 + H_2O \end{array}$$

Q.6(a) Write the structure of A and B. [2]  

$$NH_2 \xrightarrow{NaNO_2, HCI} A \xrightarrow{50\%H_2SO_4} B$$

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Q.6(c) Write the stepwise mechanism of following Fries rearrangement reaction.

$$\begin{array}{c} & & \\ & &$$

Q.7(a) Write the structure of A and B. [2]  $A \xrightarrow{C_2H_5OH} A \xrightarrow{B} B$ 

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